

Operation of Laser-Driven Light Sources below 300nm: Ozone Mitigation

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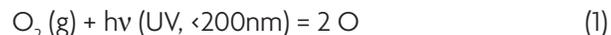
Introduction

Deep ultraviolet sources, such as Energetiq's LDLS™, can produce ozone which can adversely affect the performance of instruments and experiments connected to the LDLS. Ozone in the optical path of an instrument will absorb UV light by varying amounts, depending on the ozone concentration. This application note explains the ozone generation mechanism and appropriate ozone mitigation techniques that will ensure the best possible results.

Ultraviolet (UV) ozone generation

When air (or oxygen) is exposed to UV radiation below about 200nm wavelength, ozone will be generated. Ozone (O₃) is the tri-atomic form of molecular oxygen (O₂). It is a colorless (at low concentration, blue at high concentration), unstable gas, with a distinctive sharp odor. Ozone is considered a pollutant for human health and it irritates the throat and eyes. Good ventilation is important when ozone is produced in a working environment. Ozone is a powerful oxidizer and will oxidize and degrade organic materials with which it comes into contact.

Ozone is produced when air is exposed to deep UV radiation through dissociation of oxygen molecules. When an oxygen molecule absorbs UV radiation, it breaks apart into two oxygen atoms (Eq. 1). These oxygen atoms combine with oxygen molecules to form ozone (Eq. 2).



Ozone molecules in air can also absorb ultraviolet radiation, mainly at wavelengths between 220nm and 280nm, causing ozone molecules to decompose into oxygen molecules and oxygen atoms. (Eq. 3). Those oxygen atoms may again repeat the reaction shown in Eq. 2, or they may react with an ozone molecule to form two oxygen molecules as shown in Eq. 4.



This dynamic process can produce a situation where time-varying concentrations of ozone and oxygen are present along the UV beam path. Since both ozone and oxygen absorb deep UV light at various wavelengths, time-varying reduction in the UV is to be expected.

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Figure 1 shows the photodissociation cross sections of oxygen and ozone. Absorbance of UV radiation by oxygen and ozone is directly proportional to the photodissociation cross section as well as molecular concentration. Since the concentration of ozone in air is much lower than that of oxygen, the practical measurable absorption of ozone in air is between 220nm and 280nm, and the main absorption of oxygen in air is under 200nm. The cross section data are highly variable with temperatures and the values in Figure 1 are only indicative.

The spectral absorbance data in Figure 2 shows along a 50cm beam path without nitrogen purge using an Energetiq EQ-99 LDLS source. The peak absorbance is near 250nm, predominantly from ozone, with light being absorbed predominantly by oxygen below 200nm.

If oxygen is removed from the entire UV beam path, by purging using high purity nitrogen, then no ozone will be created and therefore the UV light will be efficiently transmitted.

Figure 1. Photodissociation cross section for oxygen molecules and ozone molecules. [1]

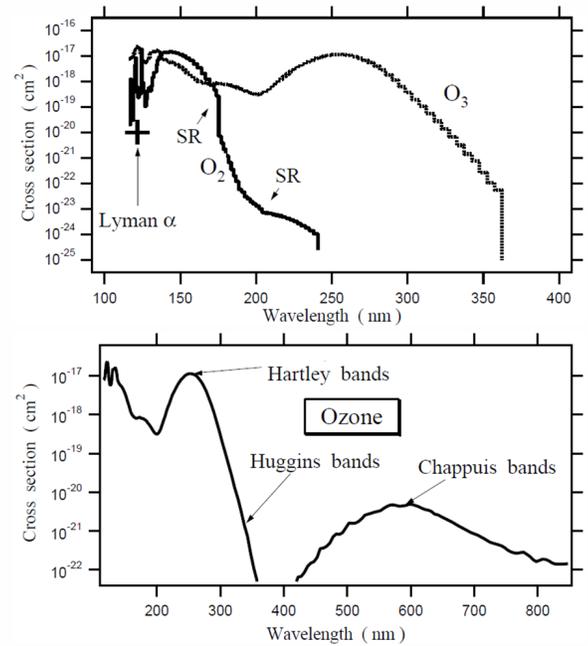
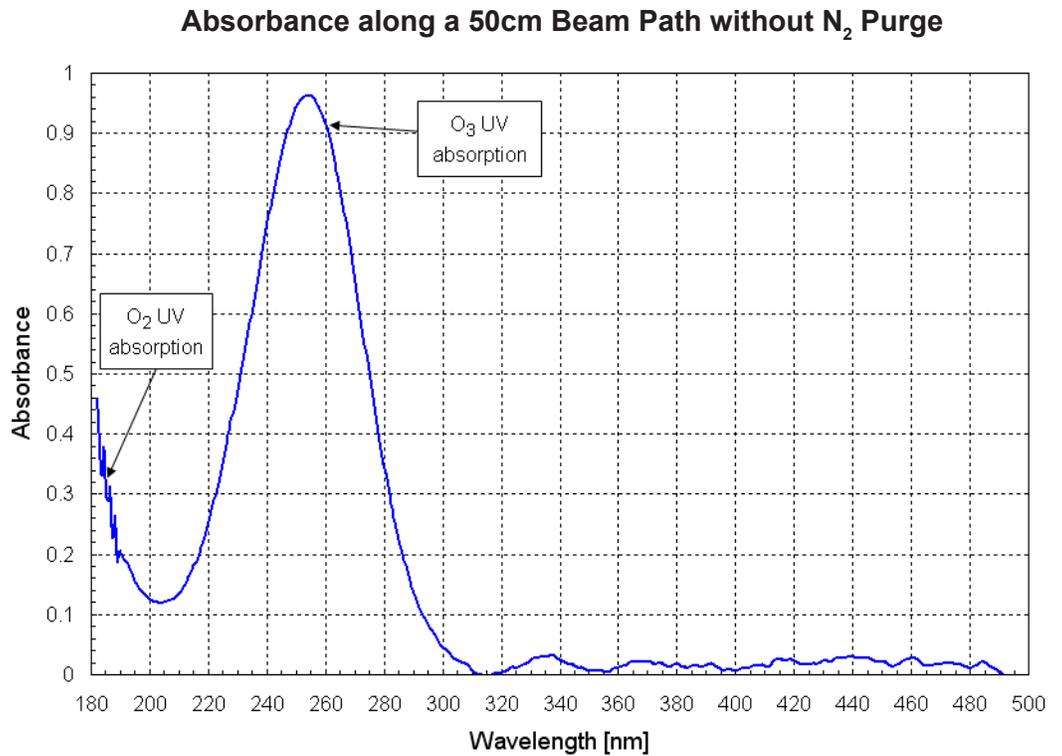


Figure 2. A spectral absorbance along a 50cm beam path, without nitrogen purge, using an EQ-99 LDLS



Ozone Mitigation in a UV optical system

Energetiq's Laser-Driven Light Sources produce high-brightness broadband radiation from 170nm (the cut off of the high quality synthetic fused silica bulb) through visible and into the near infrared. Consequently, LDLS systems will generate ozone when oxygen in air is in the beam path, both inside and outside the lamp housing. In most LDLS applications, high-purity nitrogen purging of both the lamp housing and along the entire beam path is strongly recommended. Purging the lamp house and the beam path with high-purity nitrogen has a secondary benefit. Hydrocarbons present in air can also be photodissociated by deep UV light. These photodissociated hydrocarbon fragments can deposit on surfaces in the beam path as amorphous carbon-based thin-films which reduce optical transmission and reflection of those surfaces.

A nitrogen purge port is located on all LDLS lamp houses. It is recommended that high-purity (Grade 6) nitrogen is connected to this port to mitigate ozone production within the lamp house. For the optical beam path beyond the window of the LDLS, care should be taken in the instrument design or optical bench layout to allow for purging with nitrogen. The beam path should be made with purgeable optical tubes or enclosed in a box fitted with a nitrogen purge port. Keeping the volume low for the enclosure and ensuring close fitting seals and joint will minimize the flow rate of nitrogen needed to eliminate ozone production. In cases where a tightly fitting enclosure is not practical, the beam path may be aggressively ventilated with an exhaust system. By aggressively exhausting the area around the beam path, the concentration of ozone will be reduced, and the absorption in the 250nm region will be reduced.

LDLS Applications Above 200nm

In LDLS applications where UV radiation below 200nm is not needed, special filters and window materials can be installed to replace the fused silica window on the lamp housing. Those window materials will practically eliminate the need to purge the beam path outside the lamp housing, since the major part of ozone generation is by UV radiation under 200nm.

Energetiq recommends that the lamp house still be nitrogen purged to prevent ozone buildup inside the lamp housing. Ozone in the lamp house will absorb radiation between 220nm and 280nm.

Summary

- Ozone will be generated if air is exposed to radiation at $< 200\text{nm}$
- Ozone will reduce light output at 250nm, $\pm 30\text{nm}$
- Exposure to ozone should be minimized for operator safety
- The LDLS lamp house should be purged with nitrogen. The external beam path can be purged with nitrogen or actively exhausted to prevent ozone accumulation
- For applications $> 200\text{nm}$, ozone generation in the beam path can also be prevented by using a window on the LDLS that blocks radiation $< 200\text{nm}$

References

[1] G. Kockarts, Annales Geophysicae (2002) 20: 585-598 (<http://hal-insu.archives-ouvertes.fr/docs/00/31/69/92/PDF/angeo-20-585-2002.pdf>)